

## Characterization of UV-Peroxide Killing of Bacterial Spores

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### ABSTRACT

Advantage is taken in many sterilization processes, especially for food packaging materials, of the synergy between  $H_2O_2$  and UV irradiation for spore killing. The nature of the synergy is currently not well defined in terms of targets and mechanisms. We found that under some experimental conditions, the synergistic killing of spores of *Bacillus megaterium* ATCC 19213 appeared to be mainly UV-enhanced peroxide killing, while under other conditions, it appeared to be mainly peroxide-enhanced UV killing. Lethal combinations of  $H_2O_2$  and UV irradiation for spores resulted in only modest increases in auxotrophic mutations among survivors, indicative of little DNA damage, in contrast to higher mutation levels after dry-heat damage at 115°C. However, the combination of UV light and peroxide did lead to major inactivation of glucose 6-phosphate dehydrogenase, an enzyme that was used to monitor the damage to bacterial protein. Synergistic UV- $H_2O_2$  killing was reduced by agents such as pyruvate, thiosulfate, and iron or copper cations, which appeared to act at least in part by reacting chemically with  $H_2O_2$ , and was only slightly affected by the use of UV light at a wavelength of 222 nm rather than 254 nm. Hydrogen peroxide treatment can precede UV irradiation for synergistic killing by some hours with an interim of drying for spores of *Bacillus subtilis* A, a spore type used commonly for the validation of aseptic processes. Synergistic killing of dried spores or those in suspensions was accelerated at higher temperatures (50°C) rather than at lower temperatures (25°C).

UV irradiation and hydrogen peroxide can act synergistically to kill bacteria, both vegetative cells and spores. Results of initial studies by Bayliss and Waites (2–4) indicate that the two agents had to act at the same time rather than sequentially for synergy. Advantage has been taken of this synergistic action to design more effective sterilization procedures that have been especially directed to the killing of spores. Its major application has been seen in aseptic packaging and in the processing of packaging materials undergoing peroxide sterilization, which will subsequently be used with products sterilized by ultrahigh-temperature processes. Aseptic technology is now being applied to an increasingly wide variety of products from foods to pharmaceuticals. The optimal wavelength of irradiation that will kill spores in the presence of  $H_2O_2$  is approximately 270 nm, with an effective range from about 240 to 290 nm (16). The optimal  $H_2O_2$  concentration for UV- $H_2O_2$  killing at 20°C was about 0.3 M (ca. 0.92%) for spores of *Bacillus subtilis* (2), and killing was reduced at higher concentrations, probably because of UV shielding by the peroxide. Use of 0.3 M of  $H_2O_2$  alone was not lethal for the spores. The sporicidal action of  $H_2O_2$  is very temperature sensitive (14), so that, at higher temperatures, much lower concentrations of  $H_2O_2$ , which shield less, are highly effective for killing. In essence, the shielding effect can be reduced and the sporicidal potency increased simply by raising the temperature.

Although Waites et al. (16) suggested that the mechanism for UV- $H_2O_2$  synergy is related to enhanced production of hydroxyl radicals from  $H_2O_2$  due to irradiation rather

than to any direct interactions of UV light with DNA, there remain many questions about the nature of killing by the combined agents. Is it mainly peroxide killing or UV killing, or is it somehow different from the two? What are the targets and mechanisms for killing? In this study, we have characterized spore killing by these combined agents in terms of induction of auxotrophic mutation, a sign of DNA damage, and in terms of the inactivation of a sentinel enzyme within intact spores, a sign of protein damage. We have also assessed the protective effects of transition metal cations, thiosulfate, dimethylthiourea, and pyruvate. In addition, we have investigated the feasibility of using a shorter wavelength of UV (222 nm) for spore killing and for augmenting peroxide damage.

### MATERIALS AND METHODS

**Spores.** Spores of *Bacillus megaterium* ATCC 19213 and *Clostridium sporogenes* ATCC 7955 were prepared as described previously (8). Spores of *B. subtilis* A (ATCC 9372 and NCA 7252) were prepared with the sporulation agar of Kim and Naylor (7). Spores were purified from cultures after lysis of sporangia by repeated centrifugation and the scraping of pellets to remove vegetative debris. The final preparations consisted only of phase-bright spores as viewed under a phase-contrast microscope with a 90°-positive plate. They were stored under USP 200 proof ethanol in the cold until used.

**Killing assays.** Killing assays were carried out as described previously (14) with spore suspensions containing approximately  $10^9$  CFU per ml of 1% (wt/vol) (Difco, Detroit, Mich.) peptone broth. Subsequent dilutions were in 1% Difco peptone broth, and 0.1-ml samples were streak plated on Trypticase soy agar. The addition of catalase to the peptone dilution medium did not enhance survival nor did the addition of glucose to the plating me-

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dium or the use of Plate Count Agar (Difco) instead of Trypticase soy agar. Plates were incubated for at least 48 h at 37°C to allow the full development of colonies. UV-peroxide killing was somewhat reduced, by about 50% in terms of the rate of killing, for suspensions in peptone compared with those in water or salt solution, but the final levels of killing were unaffected. Also, the killing rate could be increased somewhat by lowering the initial population levels to  $10^7$  to  $10^8$  CFU per ml, possibly because of reduced shielding. However, we preferred to maintain high population levels to more closely approximate heavily contaminated suspensions. Also, when spores were dried on glass coverslips or filter paper strips, population densities were high, even when dilute suspensions were used. The important point is that in any set of experiments, conditions of suspending medium and spore density were kept constant to keep shielding at as constant a value as possible. In fact, it is impossible to have no shielding, even with dilute spore suspensions in water. Viable counts of spores of *C. sporogenes* were determined by means of the standard most probable number assay with three tubes per dilution. All experiments were repeated at least twice. Data are presented with error bars indicating standard deviations when the number of replicates of the same experiment was three or more.

UV irradiation at 254 nm was carried out with a Sterilaire Series unit (UVP Ultra-violet Products, Cambridge, UK) with two 15-W mercury tubes, which produce mainly UV radiation at 254 nm at an intensity of 1,600  $\mu\text{W}/\text{cm}^2$  at a distance of 30.5 cm. A UVX radiometer and a sensor calibrated for radiation at 254 nm were used routinely to assess output. UV irradiation at 222 nm was carried out with an L-Bluelight Excimer Module from Heraeus Noblelight GmbH (Hanau, Germany). For irradiation, 5 ml of spore suspension was placed in a glass petri dish with a diameter of 8 cm and an area of 50  $\text{cm}^2$ . One-milliliter samples of the suspension were withdrawn at intervals during irradiation. For combined treatments, the spores were suspended at the time of irradiation in peroxide solutions. Most experiments were carried out at 25°C. The temperature was increased to as high as 50°C by use of a solid, thermostatted, heating block. For irradiation of dried spores, aliquots of suspensions were dried on standard glass coverslips (18 by 18 mm). For combined UV- $\text{H}_2\text{O}_2$  treatments, the peroxide was added to the spore suspension before drying. The results of previous studies (12) indicate that spores have a high capacity to retain peroxides during drying. UV doses were estimated by use of calibrated radiachromic strips (FWT-60-00; Far West Technology, Inc., Goleta, Calif.).

**Mutation assays.** Auxotrophic mutations were detected by replica plating survivors of UV irradiation,  $\text{H}_2\text{O}_2$  treatment, or combination treatment on defined Slepecky-Foster medium and complete Trypticase soy agar medium, as described previously (11). Spores treated with dry heat were used as positive controls for mutagenesis. All mutant colonies were subcultured on the two media to confirm that they were auxotrophic mutants able to grow on Trypticase soy but not on Slepecky-Foster agar. The levels of killing for inducing mutations ended with between 0.01 and 0.1% survival. Survivors were characterized from multiple experiments over a period of months, which allowed us the ability to test close to 2,000 colonies of survivors for UV,  $\text{H}_2\text{O}_2$ , or UV- $\text{H}_2\text{O}_2$  treatments and close to 1,000 for dry-heat treatment.

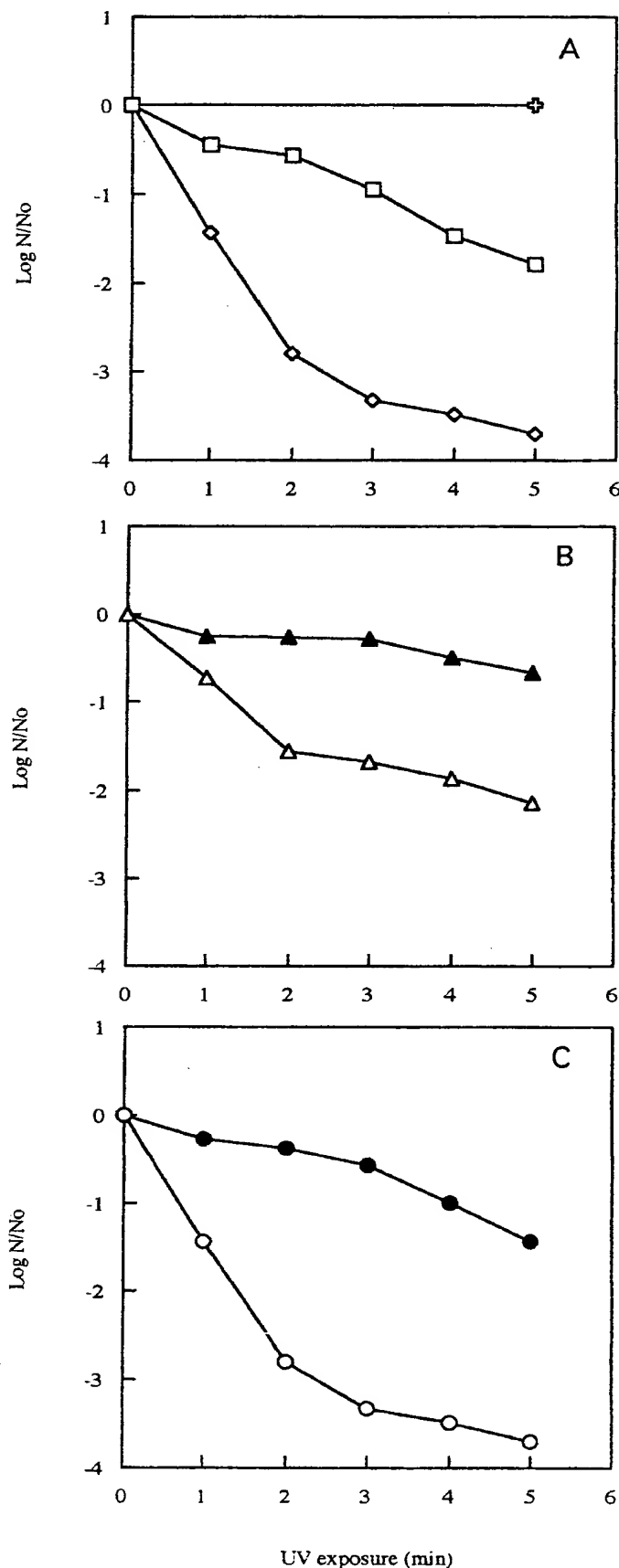
**Determination of enzyme inactivation.** Details of the assay of the enzyme glucose 6-phosphate dehydrogenase from intact spores of *B. megaterium* were given by Palop et al. (11). This assay involves monitoring changes in the absorbance of light at 340 nm because of the reduction of NADP that results from the conversion of glucose 6-phosphate to 6-phosphogluconate. Basi-

cally, intact spores were exposed to the damaging agents and then washed and germinated in the presence of chloramphenicol to block protein synthesis. The germinated spores were then permeabilized with toluene and freezing before enzyme assays. The treatments had a tendency to slow germination but did not reduce the fraction of the spores becoming phase-dark when viewed microscopically in response to germinants. Moreover, some enzymes, such as  $\text{F}_1\text{ATPases}$ , were not affected (11) by the treatments and so could not serve as indicators of UV-peroxide damage to enzymes; however, they were good indicators that the germination and permeabilization procedures were effective for treated spores.

## RESULTS

Factors affecting spore killing by UV irradiation and peroxides. The data in Figure 1A show the well-known synergistic interaction between  $\text{H}_2\text{O}_2$  and UV irradiation for spore killing, in this case, for spores of *B. megaterium* ATCC 19213 with an exposure of up to 5 min at 25°C to 0.1%  $\text{H}_2\text{O}_2$  and UV irradiation at an intensity of 0.8  $\text{mW}/\text{cm}^2$ . The concentration of  $\text{H}_2\text{O}_2$  had no effect on the viability over the short exposure. UV killing did occur with a *D*-value (time for killing of 90% of the population) of about 3 min, which equates to 1,483  $\text{J}/\text{m}^2$ . Killing with the combined agents was biphasic in this experiment, with a *D*-value of about 0.67 min for the first phase and about 6 min for the second phase. Increasing the peroxide concentration did not increase synergy, and as Waites et al. (16) found,  $\text{H}_2\text{O}_2$  had a shielding effect as well as a sensitizing effect. The synergy was specific for  $\text{H}_2\text{O}_2$  and was not found for UV light and *t*-butyl hydroperoxide or peracetic acid (12). However, this finding may be related to the use of UV light mainly at a wavelength of 254 nm, which is strongly absorbed by  $\text{H}_2\text{O}_2$  but not by the other hydroperoxides. As expected, synergy required exposure to UV light and  $\text{H}_2\text{O}_2$  at the same time with the liquid system used. Also, as found by other authors, synergy occurred only over a limited range of  $\text{H}_2\text{O}_2$  concentrations—in our experiments, from about 0.05 to 0.40%.

The results of previous studies (14, 15) indicate that transition metal cations, particularly  $\text{Fe}^{2+}$  or  $\text{Cu}^+$  but also the oxidized forms  $\text{Fe}^{3+}$  and  $\text{Cu}^{2+}$ , can protect spores against killing by  $\text{H}_2\text{O}_2$ . This same effect occurred in our studies with the combined UV- $\text{H}_2\text{O}_2$  treatment, as shown by the data presented in Figure 1B for  $\text{Fe}^{2+}$  and in Figure 1C for  $\text{Cu}^{2+}$ . The transition metal cations had essentially no effect on spore killing at 25°C by UV light alone (data not shown). The level of protection was dependent on the order of addition of the components to the mix. When  $\text{FeSO}_4$  was added to  $\text{H}_2\text{O}_2$  at the same time as or just before the addition to the spores, the salt was highly protective, but when it was added to the spores shortly after  $\text{H}_2\text{O}_2$ , the protective effect was greatly diminished or totally eliminated (data not shown). The results of previous studies (12) indicate that  $\text{H}_2\text{O}_2$  is very rapidly taken up by spores. Apparently, the peroxide within spores does not react readily with  $\text{Fe}^{2+}$  added to the suspending medium. This effect may be the basis for the importance of order of addition in relation to UV- $\text{H}_2\text{O}_2$  spore killing. If  $\text{Fe}^{2+}$  or  $\text{Cu}^+$  was added along with or before  $\text{H}_2\text{O}_2$  but prior to irradiation (Fig. 1),



killing was reduced. If the cations were added to the spores after  $\text{H}_2\text{O}_2$  and the spores were then UV irradiated, there was little reduction in killing. The simplest interpretation is that the transition metal cations rapidly catalyzed dismutation of  $\text{H}_2\text{O}_2$  to oxygen and water outside the spores. This reaction can readily be demonstrated in terms of the visible production of gas ( $\text{O}_2$ ) from mixtures containing solutions of  $\text{FeSO}_4$  and  $\text{H}_2\text{O}_2$  or the loss of  $\text{H}_2\text{O}_2$  assayed with the horseradish-peroxidase method (9).

For the experiments described in Figure 1, spore killing appeared to be predominantly by UV light, and 0.1%  $\text{H}_2\text{O}_2$  alone had only a minimal effect at  $25^\circ\text{C}$  for the short times of exposure, although it clearly enhanced killing. However, at  $50^\circ\text{C}$ ,  $\text{H}_2\text{O}_2$  was more sporicidal, and, as shown by the data presented in Figure 2, it was possible to obtain synergistic killing under conditions in which  $\text{H}_2\text{O}_2$  killing was greater than UV killing in control suspensions. Under these conditions,  $\text{Fe}^{2+}$  and  $\text{Cu}^{2+}$  were as highly protective against synergistic killing as they were at  $25^\circ\text{C}$ .

Thiosulfate at a concentration of 20 mM was at best only somewhat protective against UV- $\text{H}_2\text{O}_2$  killing at  $25^\circ\text{C}$  when UV damage was prominent, as shown by the example in Figure 3A, but was clearly more protective at  $50^\circ\text{C}$  when  $\text{H}_2\text{O}_2$  was more damaging (Fig. 3B). Dimethylthiourea, a known scavenger of the hydroxyl radical, was also protective against UV- $\text{H}_2\text{O}_2$  killing, but the protection was approximately the same at  $50^\circ\text{C}$  as at  $25^\circ\text{C}$ . For example, an average 4-log reduction in viability was found for spores of *B. megaterium* at  $25^\circ\text{C}$  after exposure to UV light at a dose of  $470 \text{ J/m}^2$  with 0.5%  $\text{H}_2\text{O}_2$ . An average of only a 1-log reduction was found for spores in the presence of 10 mM of dimethylthiourea. At  $50^\circ\text{C}$ , with a UV dose of  $470 \text{ J/m}^2$  and 0.5%  $\text{H}_2\text{O}_2$ , there was a 5-log reduction in viability without dimethylthiourea but only a 1.5-log reduction with the scavenger present.

As shown in Figure 4A, 167 mM of pyruvate was protective against UV- $\text{H}_2\text{O}_2$  spore killing as well as for killing by UV light alone. As expected, pyruvate was protective against killing by  $\text{H}_2\text{O}_2$  alone over a longer period of exposure at  $25^\circ\text{C}$  (Fig. 4B). Protection against peroxide killing presumably depends on previously documented reactions of  $\text{H}_2\text{O}_2$  with pyruvate or other  $\alpha$ -keto acids (6). In our system with 0.1% (ca. 32.6 mM)  $\text{H}_2\text{O}_2$ , degradation at  $25^\circ\text{C}$  through reactions with 167 mM of pyruvate occurred at a rate of  $0.78 \mu\text{mol/ml/min}$  when peroxide disappearance was assayed by the standard method involving the horseradish-peroxidase-leuco-crystal-violet procedure of Mottola

FIGURE 1. Killing of spores of *B. megaterium* ATCC 19213 by peroxide and UV radiation. (A) Effects of +, 0.1% (ca. 32.6 mM)  $\text{H}_2\text{O}_2$ ;  $\square$ , UV irradiation; or a combination of the two ( $\diamond$ ); (B) killing by UV light plus  $\text{H}_2\text{O}_2$  of spores suspended in 0.1%  $\text{H}_2\text{O}_2$  without ( $\triangle$ ) or with ( $\blacktriangle$ ) 10 mM of  $\text{FeSO}_4$  added just prior to irradiation; (C) killing by UV light plus  $\text{H}_2\text{O}_2$  of spores suspended in 0.1%  $\text{H}_2\text{O}_2$  without ( $\circ$ ) or with ( $\bullet$ ) 10 mM of  $\text{CuSO}_4$  solution just prior to irradiation. One minute of irradiation corresponds to a UV dose of  $480 \text{ J/m}^2$ .

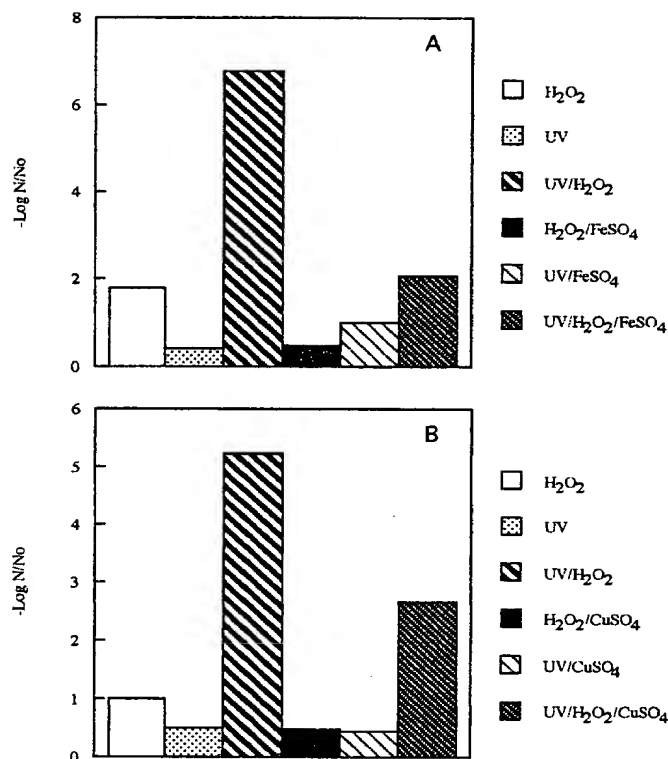


FIGURE 2. Killing of spores of *B. megaterium* ATCC 19213 at 50°C by 0.15% H₂O₂ for 2 min and/or UV irradiation for 2 min at 3.5  $\mu\text{W}/\text{cm}^2$  for 2 min (4.2 J/m²) with or without 10 mM of FeSO₄ (A) or CuSO₄ (B).

et al. (9). This rate is somewhat lower than the rate of about 2.5  $\mu\text{mol}/\text{ml}/\text{min}$  reported by Giandomenico et al. (6) for a mixture of 1 mM of pyruvate and 200  $\mu\text{M}$  of H₂O₂ in water for 1 h at 37°C. The finding that pyruvate can protect against UV spore killing alone was surprising. Protection against UV killing presumably does not depend on peroxide degradation unless, of course, UV irradiation results in the formation of peroxides that can react with pyruvate. Still, pyruvate protection against both of the individual agents has to be considered in interpreting protection against the combined agents. At 50°C, pyruvate was highly protective against UV-H₂O₂ killing. For example, there was a 6.7-log reduction in the viable count of spores subjected to 0.15% H₂O₂ at 50°C for 2 min with UV irradiation at 3.5  $\mu\text{W}/\text{cm}^2$  (4.2 J/m²). Only a 0.3-log reduction in the viable count occurred when 167 mM of pyruvate was added to the spore suspension just prior to treatment with UV light and peroxide.

**Use of UV irradiation at 222 nm.** A shorter wavelength (222 nm) of UV irradiation proved to be more effective than the standard UV light from a mercury lamp (ca. wavelength of 254 nm) commonly used for spore killing. However, it was only somewhat more effective for killing spores either in aqueous suspensions or dried on glass, as shown by the sample data in Figure 5. Sample data are presented for *B. megaterium* spores dried on glass coverslips (Fig. 5A) and for *C. sporogenes* spores (Fig. 5B) in liquid suspensions.

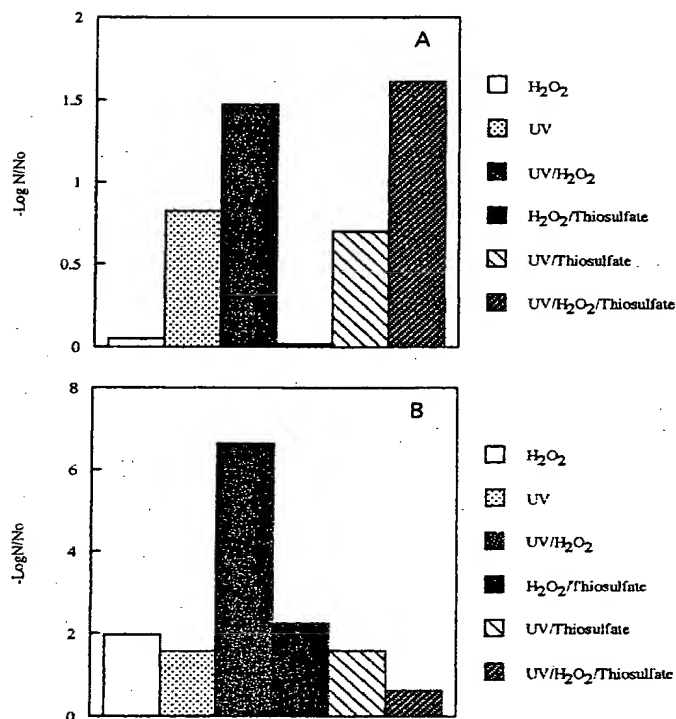


FIGURE 3. Effects of 20 mM of sodium thiosulfate on killing of spores of *B. megaterium* ATCC 19213 at pH 7 and 25°C by UV irradiation at an intensity of 1.15 mW/cm² for 45 s (518 J/m²), by 0.1% H₂O₂ for 45 s, or by a combination of the two (A). Data are shown in (B) for the effects of 20 mM of sodium thiosulfate on killing at 50°C by 0.15% H₂O₂ for 2 min, UV irradiation at 3.5  $\mu\text{W}\cdot\text{cm}^{-2}$  for 2 min (4.2 J/m²), or both.

As shown by the data in Figure 6, the shorter wavelength of UV light was effective for synergistic killing of spores of *B. subtilis* A, which is a spore type commonly used to assess the effectiveness of aseptic processing. The synergy for spores in suspensions is obvious from the data in Figure 6A. Also, we undertook experiments of the type described previously (12) for spores of *B. megaterium* or *C. sporogenes*; however, in our study, we used spores of *B. subtilis* A and with a UV wavelength of 222 nm. Spores of *B. subtilis* A were suspended in a solution containing 0.65% H₂O₂, and aliquots of the suspension were dried on glass coverslips at 25°C for various times before irradiation. Spores of *B. megaterium* and *C. sporogenes* can retain H₂O₂ under these drying conditions for as long as 24 h, although the peroxide completely evaporates from coverslips without spores (12). The data presented in Figure 6B through D indicate that spores of *B. subtilis* A must also retain peroxide because, even after 6 or 12 h of drying, synergistic killing was evident. Even after 24 h of drying, the agents still appeared to have at least additive action. However, more prolonged drying resulted in further reductions in peroxide action. Similar results were obtained with spores of *B. megaterium* and *C. sporogenes* when UV light at wavelength 222 rather than 254 nm was used for the killing of cells previously exposed to H₂O₂ and then dried on coverslips for various times. Differences in the effectiveness of UV light at 222 versus 254 nm for synergistic

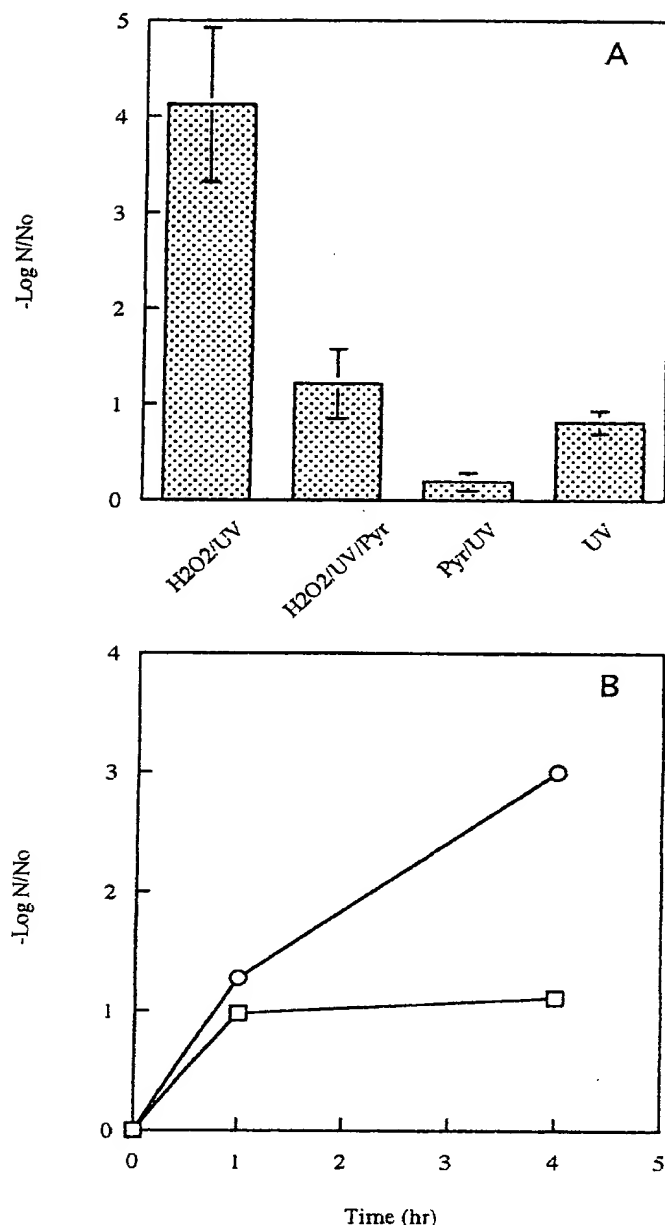


FIGURE 4. Pyruvate inhibition of killing of spores of *B. megaterium* ATCC 19213. (A) Data for effects of 167 mM of sodium pyruvate on spore killing by UV irradiation at an intensity of 1.52 mW/cm<sup>2</sup> for 60 s (912 J/m<sup>2</sup>) at 25°C and of UV irradiation of spores in 0.1% H<sub>2</sub>O<sub>2</sub> solution—bars show standard deviations with  $n = 3$ ; (B) data that show inhibition by 167 mM of sodium pyruvate (□) on killing of spores by 0.5% H<sub>2</sub>O<sub>2</sub> alone (○) at 25°C and pH 7 during a 4-h period.

killing were too small for reliable distinctions to be made between the two (data not shown). However, UV light at 222 nm appears to be as effective as at 254 nm for synergistic killing.

#### Nature of the damage caused by UV light and H<sub>2</sub>O<sub>2</sub>.

As shown by the data in Figure 7, auxotrophic mutants were obtained readily after dry heating. Spores of *B. megaterium* ATCC 19213, the strain used for most of our previous work on synergistic killing, were dried at room tem-

perature on glass coverslips and then heated at 115°C for 5 min. About 2% of the survivors were able to grow in complex medium but not on defined Slepecky-Foster medium. The response to UV irradiation (270 J/m<sup>2</sup>) was less mutagenic, and fewer than 1% of the survivors were unable to grow on the defined medium. Combined UV-H<sub>2</sub>O<sub>2</sub> treatment was even less mutagenic, while treatment with 0.1% H<sub>2</sub>O<sub>2</sub> resulted in only very low levels of auxotrophic mutation. The treatment times varied from 5 min for dry heat to hours for H<sub>2</sub>O<sub>2</sub> treatment at 25°C to achieve the desired levels of spore killing. Survivors were initially plated on Trypticase soy agar with incubation for at least 48 h to allow for the full growth of colonies. Then, colonies were picked randomly for testing. Dry-heat treatment was the positive control in the experiments. No treatment resulted in no detected auxotrophic mutations but, of course, also no killing. The differences among UV treatment, H<sub>2</sub>O<sub>2</sub> treatment, and combined treatment were not statistically significant at the 95% confidence limit, although treatment with H<sub>2</sub>O<sub>2</sub> alone appeared to be the least mutagenic. The conclusion is that combined UV-H<sub>2</sub>O<sub>2</sub> treatment is not highly mutagenic for *B. megaterium* ATCC 19213 spores, at least not compared with the dry-heat treatment. The levels of auxotrophic mutation presented in this study for spores treated with peroxide are somewhat higher than those of, say, Fairhead et al. (5), but the organisms used were different, *B. subtilis* versus *B. megaterium*.

Results of our previous studies of spore killing by hydroperoxides and by heat have suggested that the inactivation of enzymes in the spore core may be involved in killing them (11). There is a wide spectrum of sensitivity among spore enzymes. For example, glucose 6-phosphate dehydrogenase in intact spores is highly sensitive to inactivation by H<sub>2</sub>O<sub>2</sub>, whereas F<sub>1</sub>ATPase has very low sensitivity. As shown by the data in Figure 8, glucose 6-phosphate dehydrogenase was sensitive to inactivation by combined UV-H<sub>2</sub>O<sub>2</sub> treatment. Under the experimental conditions, most of the inactivation appeared to be due to peroxide damage enhanced by the low level of UV irradiation used. However, glucose 6-phosphate dehydrogenase within spores could also be inactivated by UV irradiation alone.

## DISCUSSION

The data presented suggest that the nature of combined UV-H<sub>2</sub>O<sub>2</sub> killing can be affected by experimental conditions ranging from a process dominated by UV killing to one dominated by peroxide killing. In the sterilization of packaging materials, the killing is probably mainly dominated by peroxide because the materials are exposed to high concentrations of H<sub>2</sub>O<sub>2</sub> at high temperatures before being UV irradiated. Thus, industrial conditions are closer in terms of temperature to our 50°C conditions under which thiosulfate is highly protective against killing. Of course, the industrial conditions involve much higher peroxide concentrations than we used and more rapid killing. Thus, our conditions only approach those of industrial units. Spores can take up very high concentrations of H<sub>2</sub>O<sub>2</sub>, even when exposed to fairly low concentrations of the agent (12).

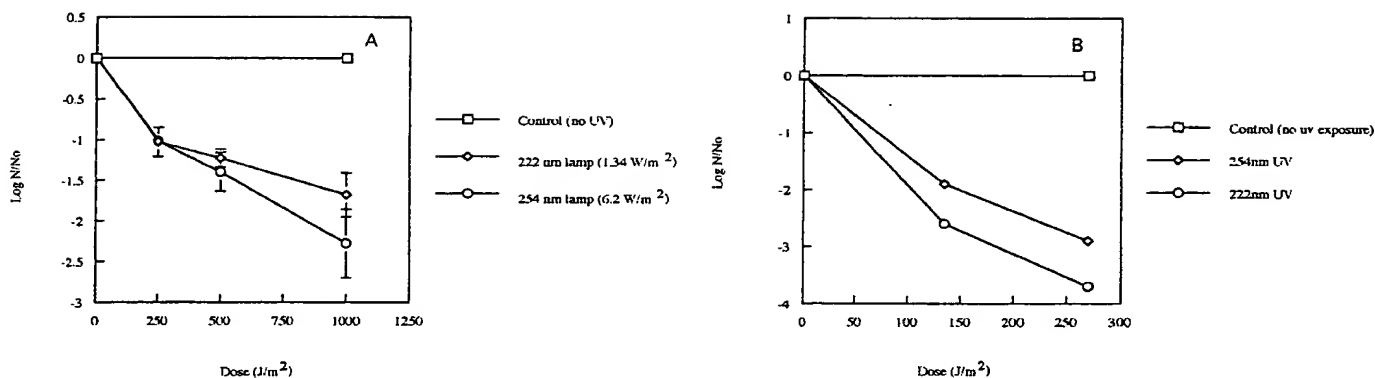


FIGURE 5. Examples of relative sporicidal effectiveness of UV irradiation at 222 compared with UV irradiation at 254 nm for dried spores of *B. megaterium* (A) or for spores of *C. sporogenes* (B) in suspensions.

Therefore, even if there is drying between peroxide exposure and UV irradiation, synergistic killing can still occur.

Fe or Cu cations or pyruvate are known protectors of spores against H<sub>2</sub>O<sub>2</sub> damage, although Fe or Cu cations are well known to act oppositely to sensitize vegetative cells to peroxide killing. The findings presented indicate that they can also protect against combined UV-H<sub>2</sub>O<sub>2</sub> spore killing. The agents appear to act primarily by chemically catalyzing H<sub>2</sub>O<sub>2</sub> dismutation outside the spore or at least external to targets for damage.

UV light or H<sub>2</sub>O<sub>2</sub>, separately or in combination, did not appear to be highly mutagenic, in contrast to dry heat

or ionizing radiation. The findings for the separate agents are consistent with those of other studies (10, 13). These findings are generally explained as being the result of protection of the spore DNA against UV light or peroxide damage by small, acid-soluble spore proteins, with lesser contributions attributed to the physical state of the spore DNA and to DNA repair enzymes active after germination. It seems the same protective mechanisms apply to combined UV-H<sub>2</sub>O<sub>2</sub> killing. Bayliss et al. (1) found that DNA repair-deficient spores had heightened sensitivities to the combined agents.

The finding that UV irradiation at 222 nm is as effec-

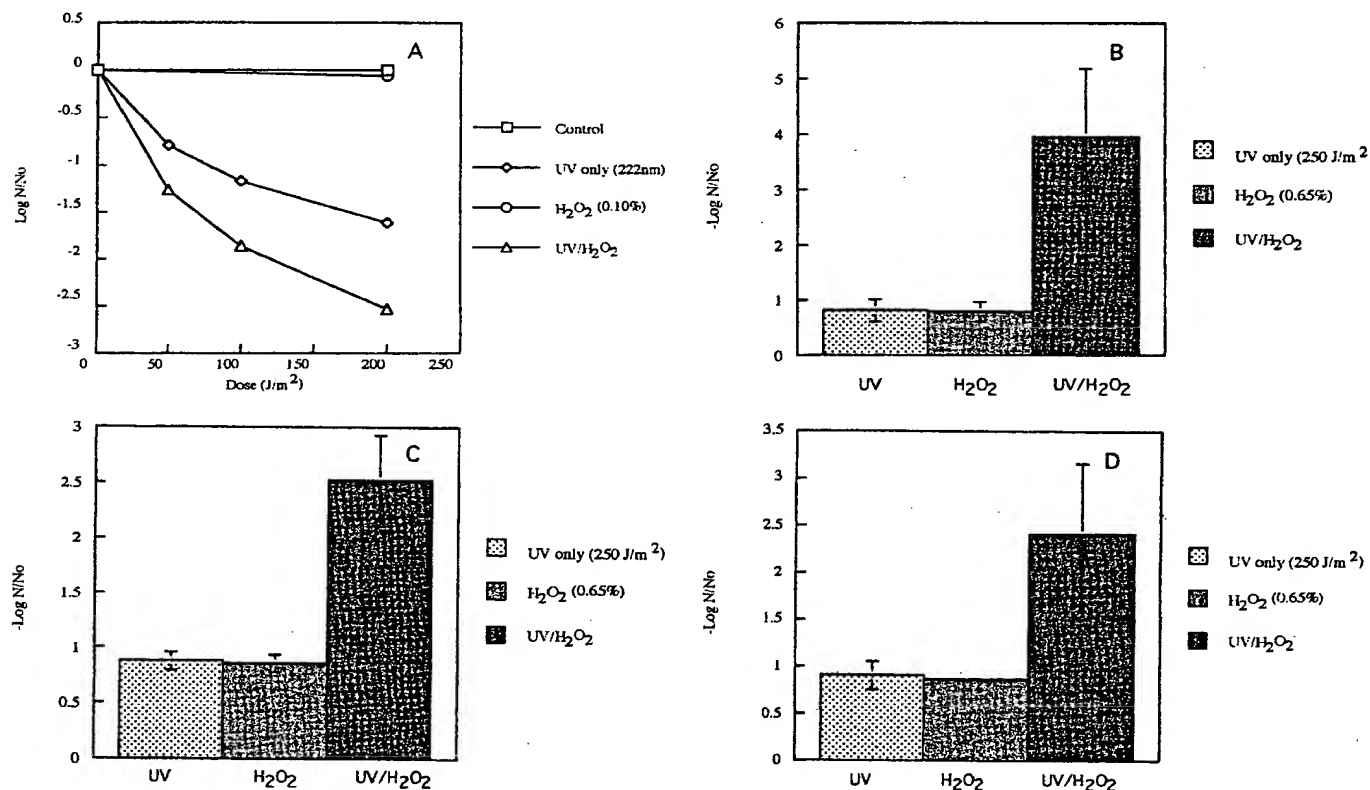


FIGURE 6. (A) Synergistic killing of *B. subtilis* A spores suspended in 0.10% H<sub>2</sub>O<sub>2</sub> solution at 25°C and then UV irradiated at 222 nm. Synergistic killing as assessed also for spores of *B. subtilis* A exposed initially to 0.65% H<sub>2</sub>O<sub>2</sub> solution and then dried on glass coverslips at 25°C for 6 (B), 12 (C), or 24 (D) h before being UV irradiated at 222 nm.

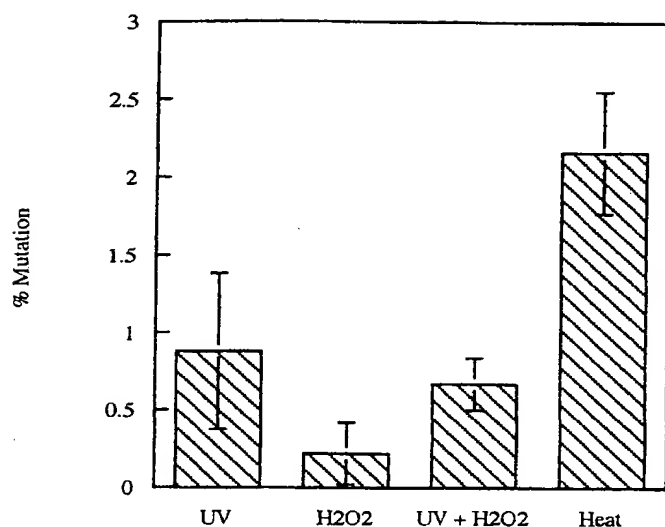


FIGURE 7. Auxotrophic mutants among survivors of killing of spore populations by UV irradiation (270 J/m<sup>2</sup>), 0.1% H<sub>2</sub>O<sub>2</sub> at 25°C, pH 7, UV irradiation of spores in 0.1% H<sub>2</sub>O<sub>2</sub> solution, or heating at 115°C of spores previously dried on glass coverslips. The error bars indicate standard deviations among multiple experiments with  $n > 3$ .

tive as UV irradiation at 254 nm does not agree with previous findings (16) obtained with spores of *B. subtilis* NCDO 2129 suspended in 1% H<sub>2</sub>O<sub>2</sub> solution and irradiated with UV light from wavelengths of 240 to 300 nm produced by a synchrotron radiation source. The basis for the difference is not clear, but it may be due to differences in experimental conditions. Also, in the previous work, the shortest wavelength used was 240 nm, and possibly, there is a trough in the curve relating sporicidal action to wavelength with a rise at 222 nm. Another possible difference is that we used more concentrated spore suspensions, about 10<sup>9</sup> versus about 10<sup>7</sup> spores per ml. However, as indicated above, in our experience, lowering the concentration of spores only somewhat enhances effectiveness of the agents but does not preclude synergy.

Our data indicate that proteins are targets for damage by the combined agents, as they are targets for damage by peroxides. We focused on glucose 6-phosphate dehydrogenase because of its known high sensitivity to peroxide damage (11). Some spore enzymes are relatively insensitive to peroxide damage, e.g., F<sub>1</sub>ATPase and pyruvate kinase. We found in separate experiments that both enzymes are also insensitive to combined peroxide-UV light damage, at least compared with glucose 6-phosphate dehydrogenase. The results of our findings with respect to the inactivation of glucose 6-phosphate dehydrogenase by the combined agents did not show major synergy, although inactivation by the combined agents was always greater than inactivation by any one of the agents. Overall, it seems that the inactivation of enzymes such as glucose 6-phosphate dehydrogenase by the combined agents could affect viability. However, cumulative damage to multiple enzyme targets is likely to be required for killing. In other words, there are multiple targets for the agents, and synergy is likely to arise

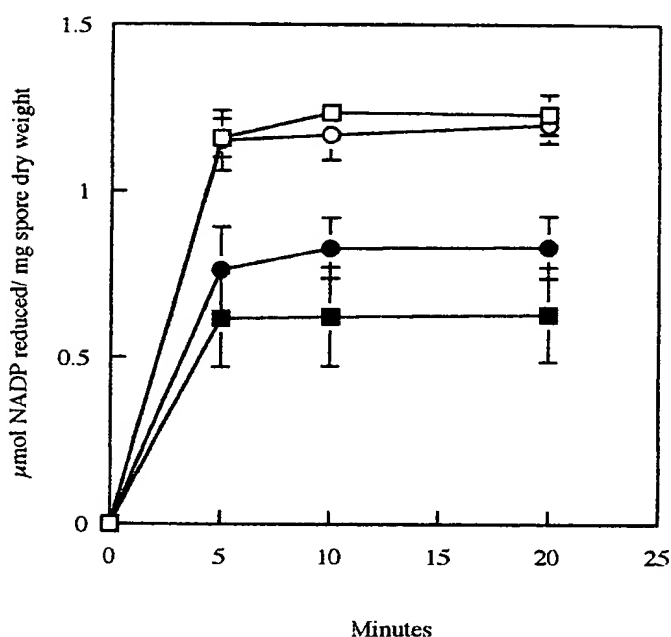


FIGURE 8. Glucose 6-phosphate dehydrogenase activities of spores after 60 min of no treatment (□), UV treatment with 3.5 μW/cm<sup>2</sup> (126 J/m<sup>2</sup>) (○), treatment with 0.1% H<sub>2</sub>O<sub>2</sub> at 25°C, pH 7 (●), or treatment with both agents (■). Error bars indicate 95% confidence limits with three trials.

from enhanced cumulative damage to multiple protein targets rather than to a single target.

Regardless of the targets for killing, it is apparent from this work as well as from previous studies (12) that synergistic killing occurs for a wide variety of spores and that exposure of spores to H<sub>2</sub>O<sub>2</sub> can precede UV irradiation by many hours when spores are dried at room temperature or for shorter times when drying is at higher temperatures. UV irradiation at 222 nm was somewhat more effective for synergistic killing than UV irradiation at 254 nm, as it is for the killing of spores in suspensions or those dried on glass coverslips. The new findings should lead to more effective use of peroxide-UV light regimens for sterilization. Certainly, they allow more freedom in machine design.

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